

# Pharmacophore-Based Techniques For the Construction of Biochemical Reaction Networks

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# Objectives

Using Automated methods to construct reaction networks, searching for **new** reactions.

Generating new pathways through large reaction networks

Obtaining kinetic expressions and fitting parameters using reduced models

# Applications

- Improving Drug Design Options
- Control Flux Analysis
- Improving Yields in Bioreactors



- Designing New Processes



- Determine the effect of New drugs
- Explore Genetic Modification and Diseases
- Adding New Capabilities to a Process

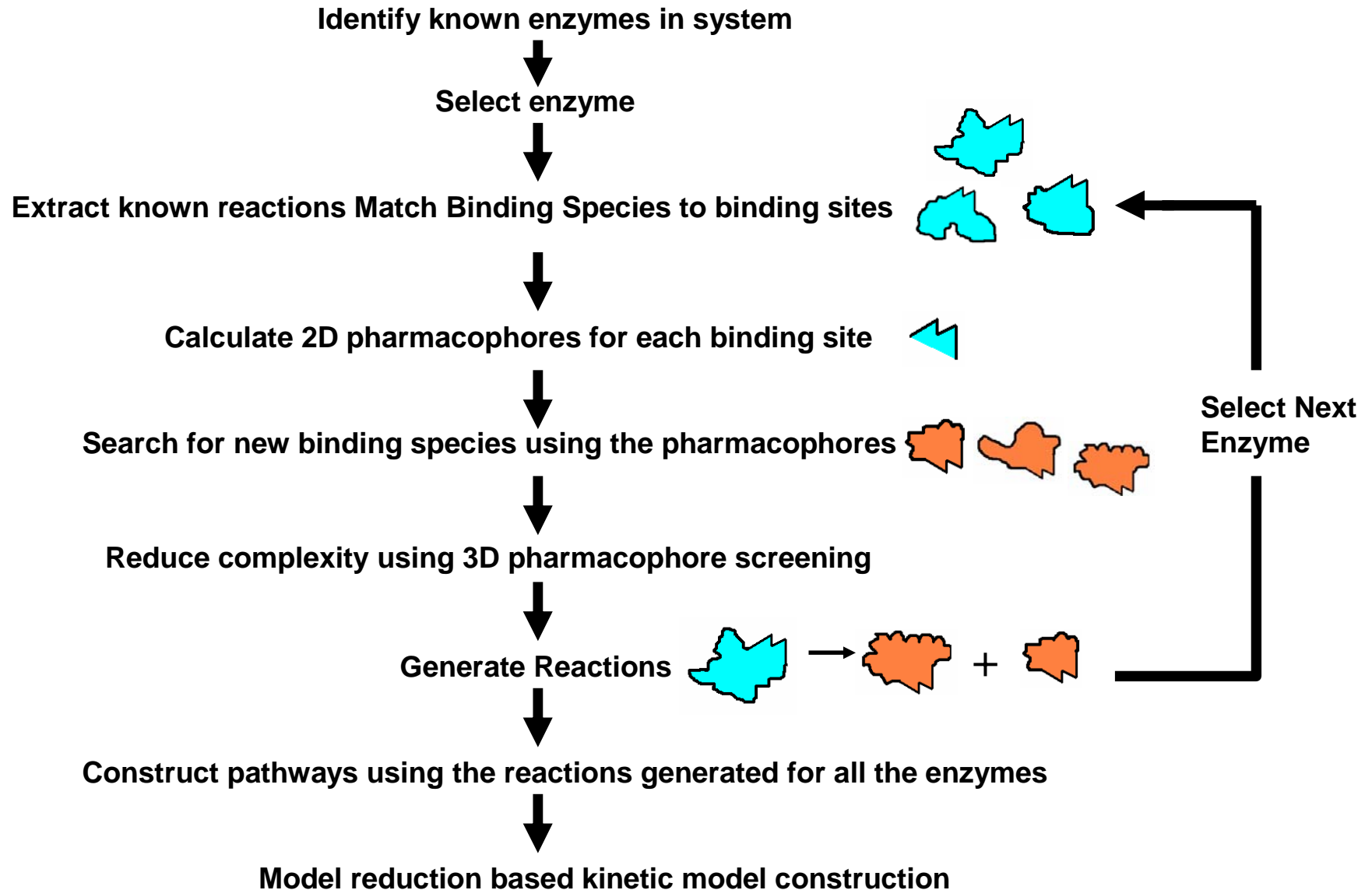


# Metabolic Network Development: Main Ideas



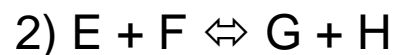
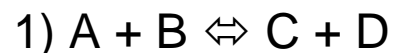
- **Initially assume that all enzymes are known**
- **Treat enzymes as black boxes**
- **Compare ligands which bind to the same sites**
- **Use this information to predict**
  - **new binding species**
  - **new reactions**

# Metabolic Network Development: A knowledge-based method for generating Biochemical reaction networks.

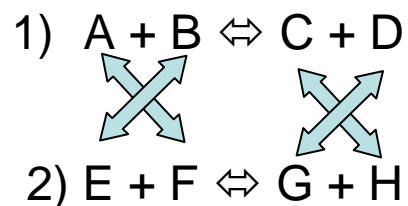
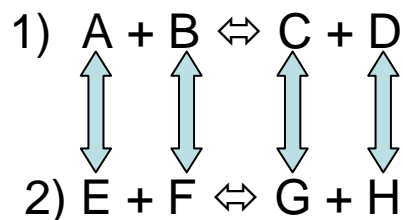


## Matching Binding Species to Binding Sites

An example enzyme catalyses the following reactions



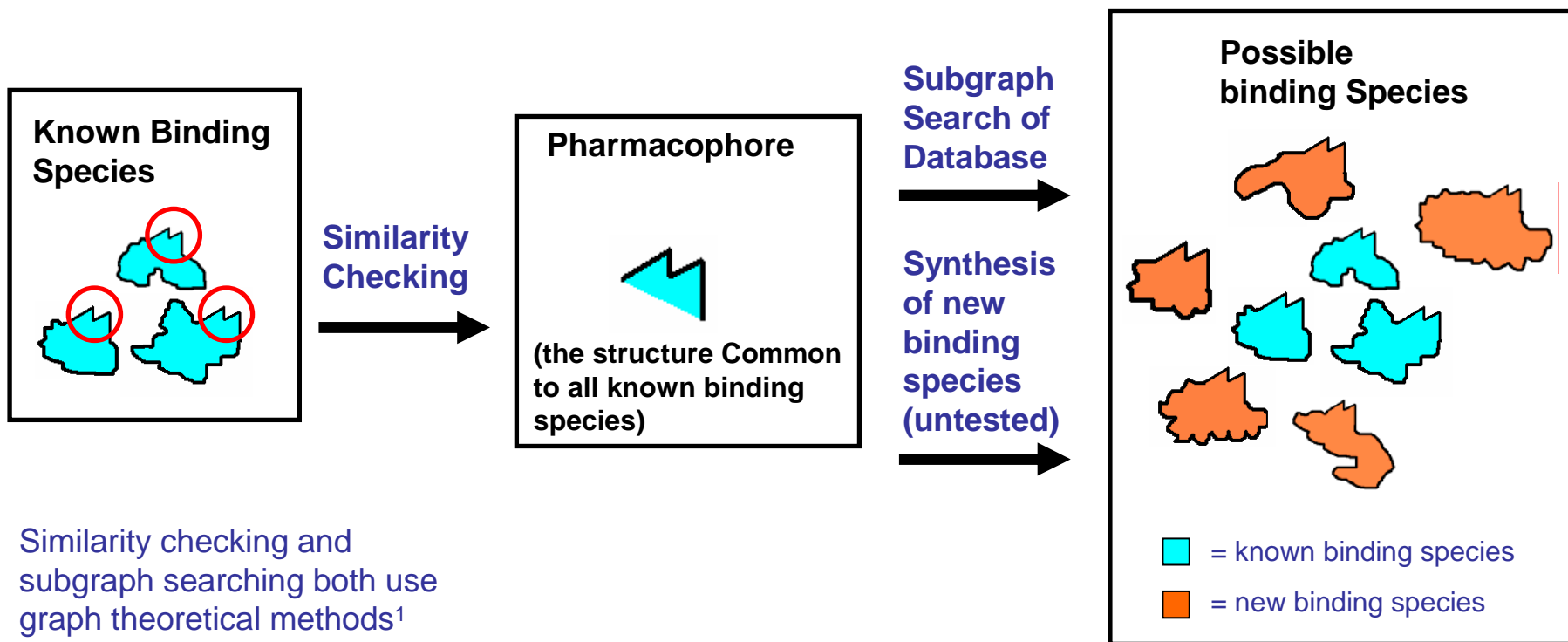
A comparison (Using Simcomp<sup>1</sup>) of the species in reaction 1 with those in reaction 2 should reveal which species bind to which sites.



Assuming that species which bind to the same site are similar

1. Hattori et. al., Journal of the American chemical society, 125: 11853-11865 (2003).

# Computation of 2D Pharmacophores



Similarity checking and subgraph searching both use graph theoretical methods<sup>1</sup> where

functional groups = nodes  
bonds = edges.

1. Hattori et. al., Journal of the American chemical society, 125: 11853-11865 (2003).

# Graph Theory

We have used the graph theory comparison methods in the program `simcomp`<sup>1</sup> to compute similarity.

Pharmacophores are computed by finding the maximal common subgraphs between binding species.



Searching for binding species is possible using a subgraph isomorphism algorithm which searches for 1 graph inside another



1. Hattori et. al., Journal of the American chemical society, 125: 11853-11865 (2003).



## Example using the enzyme 1.1.1.27

This enzyme catalyses 3 known reactions



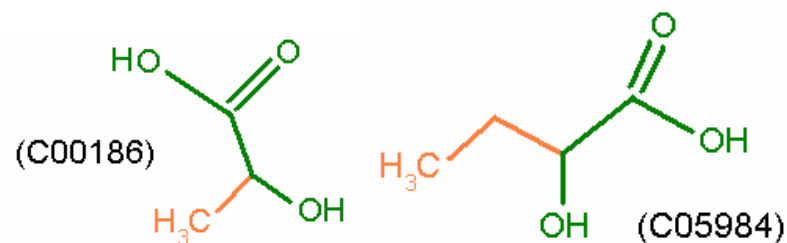
In this case only two pairs of species need to be compared.

**All Calculations were performed using raw data from the KEGG: Ligand databases<sup>2</sup> which contain biochemical reactions and the 2D structure of the compounds involved**

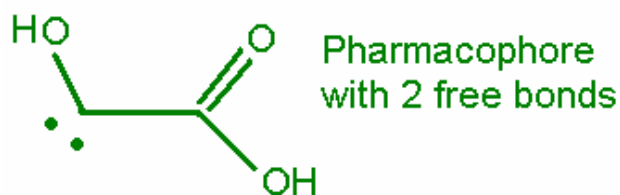
2. Goto et. al., Nucleic Acids Research, 30: 402-404, (2002).

# Comparing the reactants and products

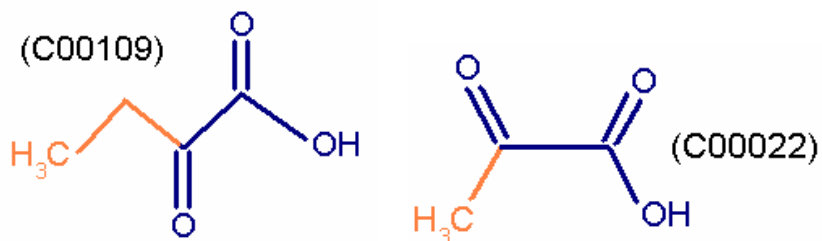
Reactants



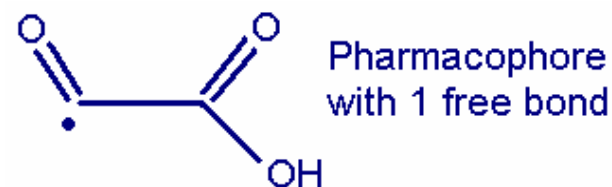
Comparison



Products

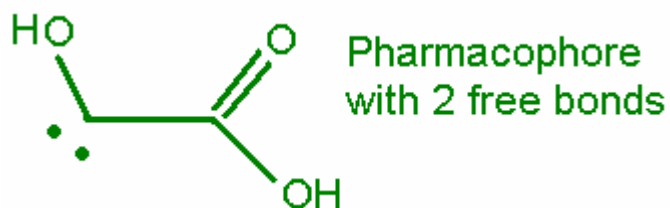


Comparison

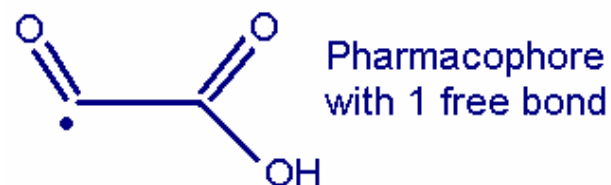


# Searching for binding species containing the 2D pharmacophores

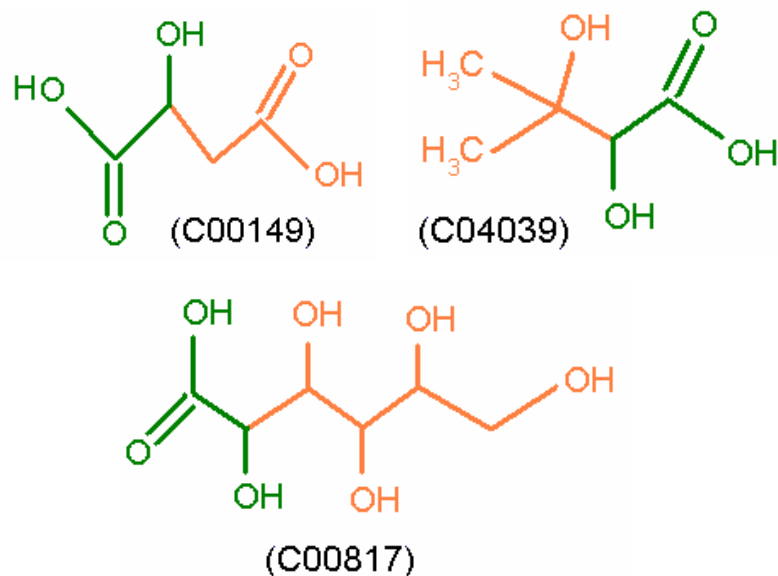
Reactants



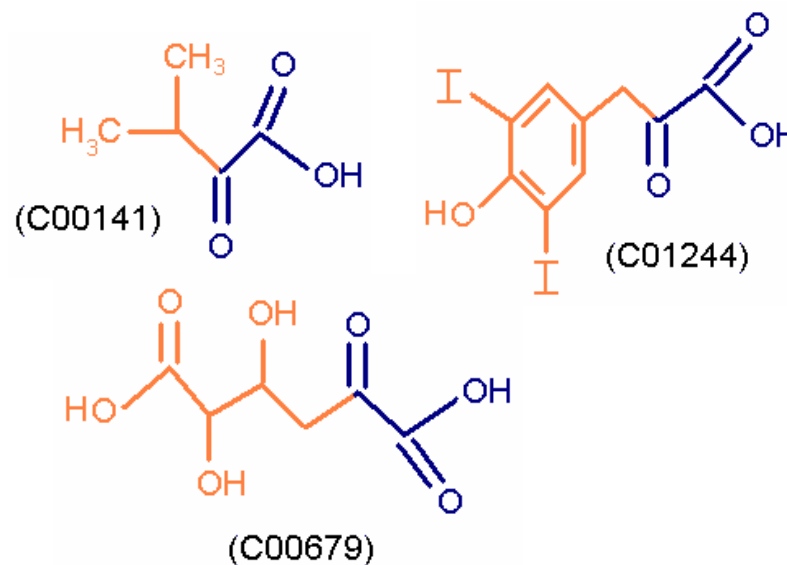
Products



118 possible binding species



139 possible binding species



# Computing 3D Structures

- Using an optimization algorithm together with an expert system
- The approximate bond lengths are found from the covalent radii of the atoms
  - With the 1<sup>st</sup> atom being placed at coordinates (0,0,0)
- All subsequent atom positions are found through the bond angles of each bond
  - Initially these bond angles are randomly generated
- The bond angles are then optimized using the following objective function
  - The 1<sup>st</sup> part of this function ensures that the correct bond lengths are maintained
  - The 2<sup>nd</sup> part maximises the distance between non-bonded pairs to avoid overlap

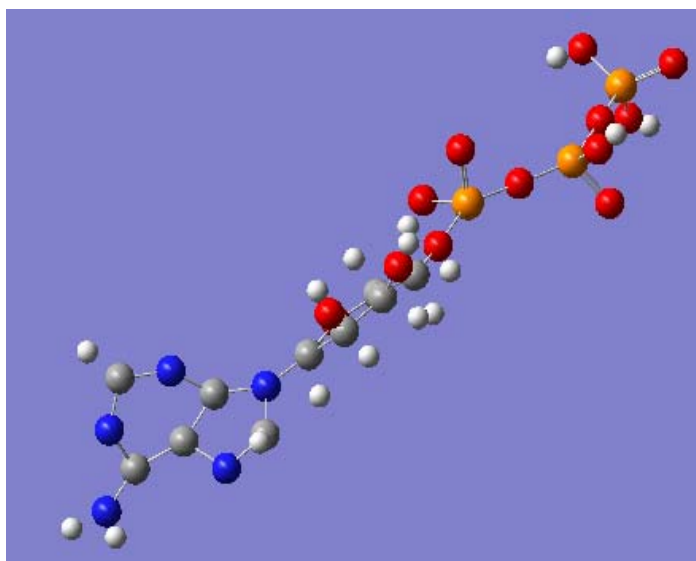
$$\min \sum_{\text{bonded}} (d_{i,j} - d_{i,j}^{\text{ex}})^2 + \sum_{\text{non-bonded}} e^{2 \times d_{i,j}^{\text{ex}} / d_{i,j}}$$

$d_{i,j}^{\text{ex}}$  Expected distances between atoms if they were bonded

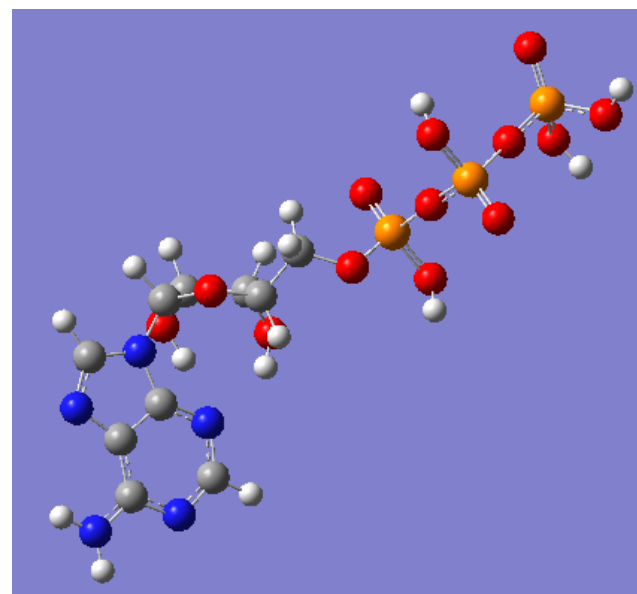
$d_{i,j}$  Actual distances between the atoms

# Structure Optimization and Charge

Approximate 3D structures can be optimized and charges can be calculated using the quantum chemical software Gaussian<sup>5</sup>. Here we have used the semi-empirical method AM1 with basis set STO-3G.



Approximate 3D Structure

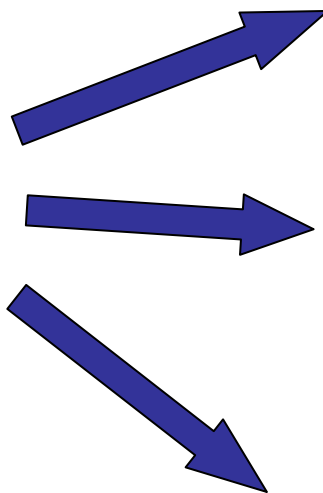
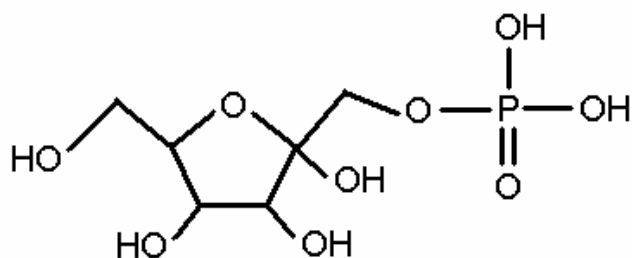


Exact 3D structure

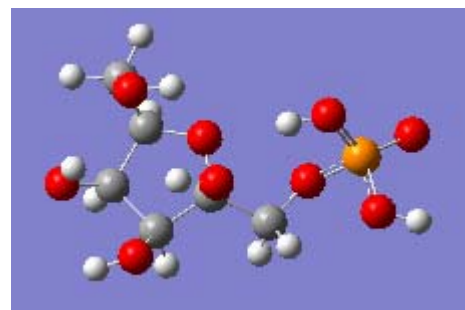
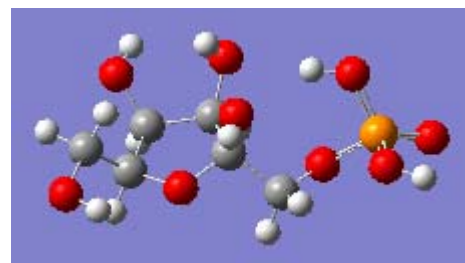
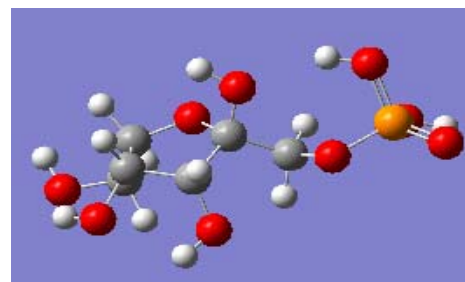
3. Frisch, M., et. Al., Gaussian-03, Inc., Wallingford CT, (2004).

# 3D Pharmacophore Generation

One 2D Structure



Many 3D conformations

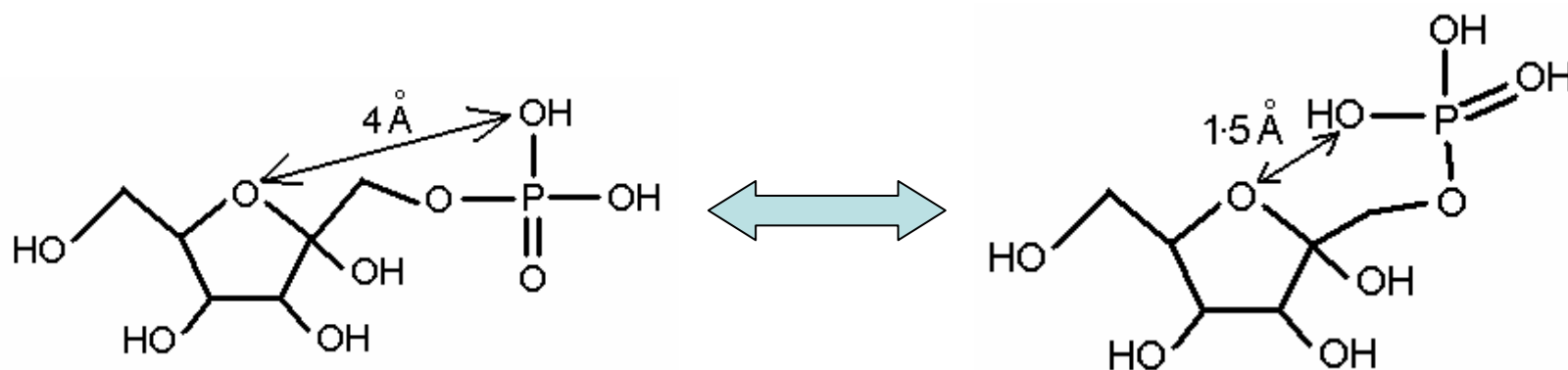


3D structures obtained using Gaussian<sup>3</sup>

3. Frisch, M., et. Al., Gaussian-03, Inc., Wallingford CT, (2004).

## Flexible 3D Bounds: A Model Reduction Step

Molecules can bend and twist



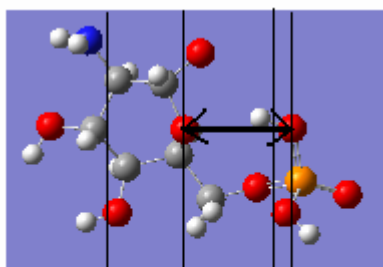
Giving a range of distances between each atom-atom pair

e.g. Oxygen-Oxygen distance = 1.5 – 4 angstroms

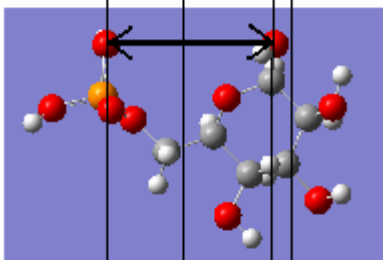
A method called Bound Smoothing<sup>4</sup> can be used to obtain these distance ranges

4. Raymond, J., Willett, P., Journal of Chemical Information and Computer Sciences, 44:908-916, (2003).

## 3D Comparisons

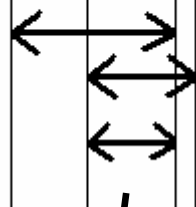


**Matching flexible structures have overlapping atom-atom distances**



Comparing of 2 molecules

Gives a smaller bound in the pharmacophore



New Binding species must be able to squeeze in this gap



## 3D Molecule Checks: Computational Requirements

A 3D graph-theoretical method is used based on atom-atom distance specification



**A typical molecule**

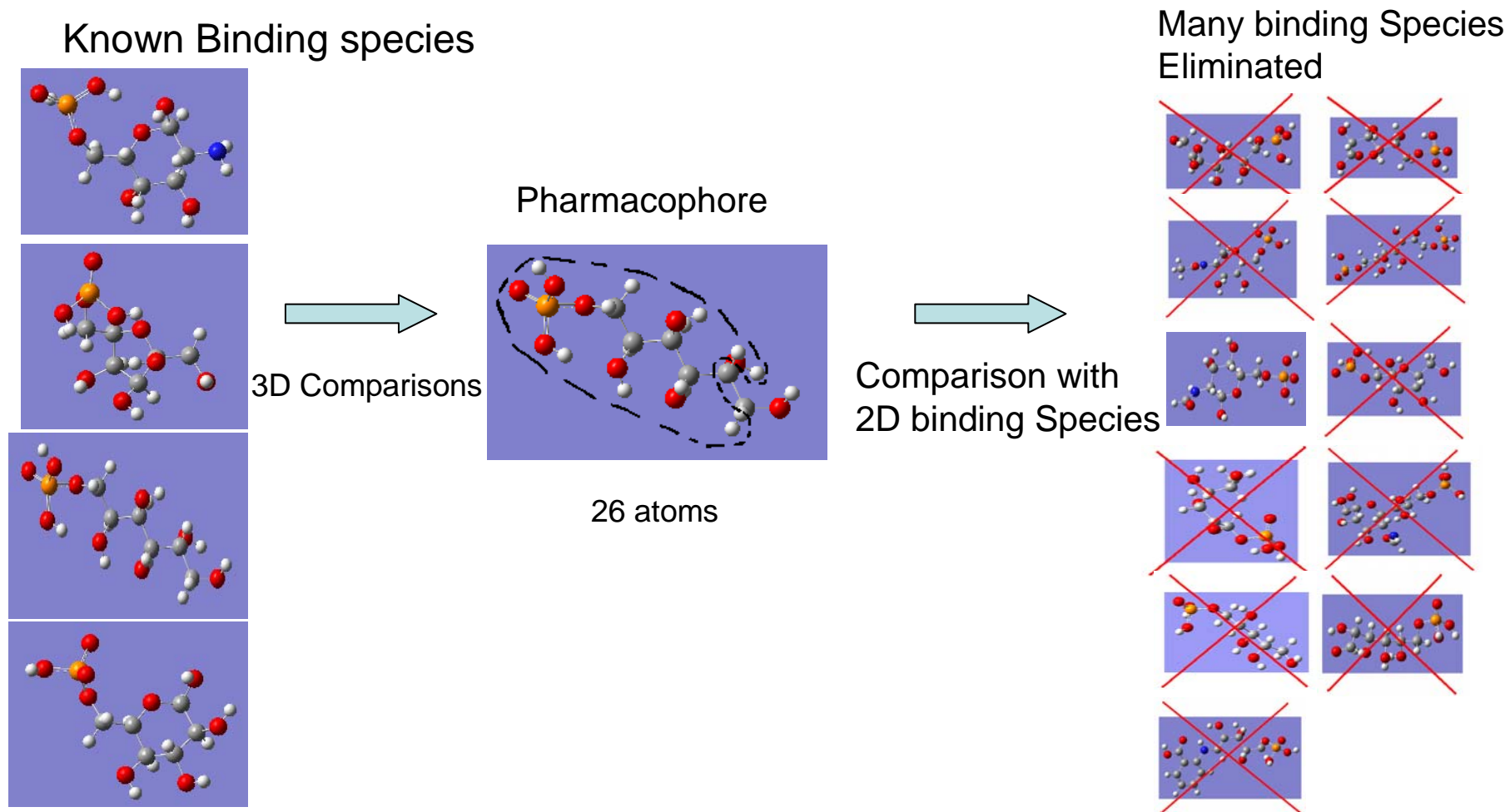
- 30 atoms**
- 30 bonds**
- 435 atom-atom distances**

Where a 2D method would use 30 bonds, the 3D method uses 435 atom-atom distances

Hence the 3D method is much more expensive

## Example: 3D Pharmacophore

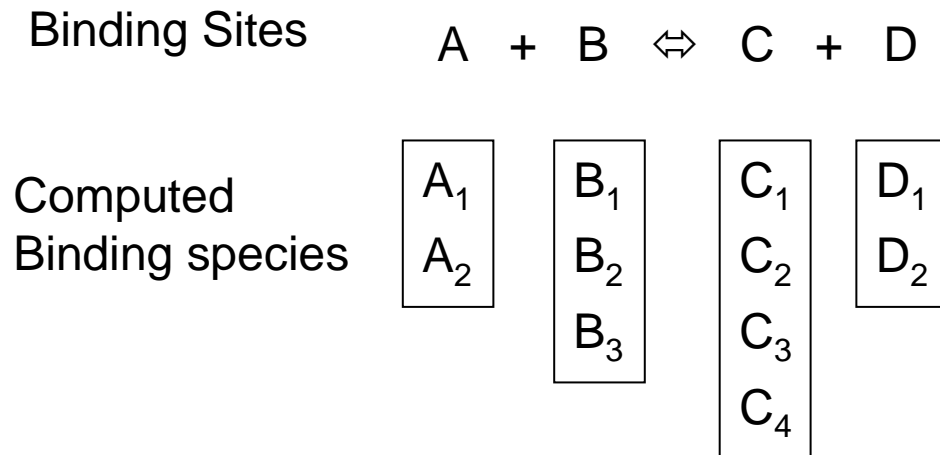
A 3D graph theoretical method<sup>4</sup> is used to eliminate 2D binding species



4. Raymond, J., Willett, P., Journal of Chemical Information and Computer Sciences, 44:908-916, (2003).

# Reaction Generation

For each enzyme generate combinations of binding species



Combinations of binding species which can form reactions

- Contain no more than 1 species from each binding site on each side of the reaction

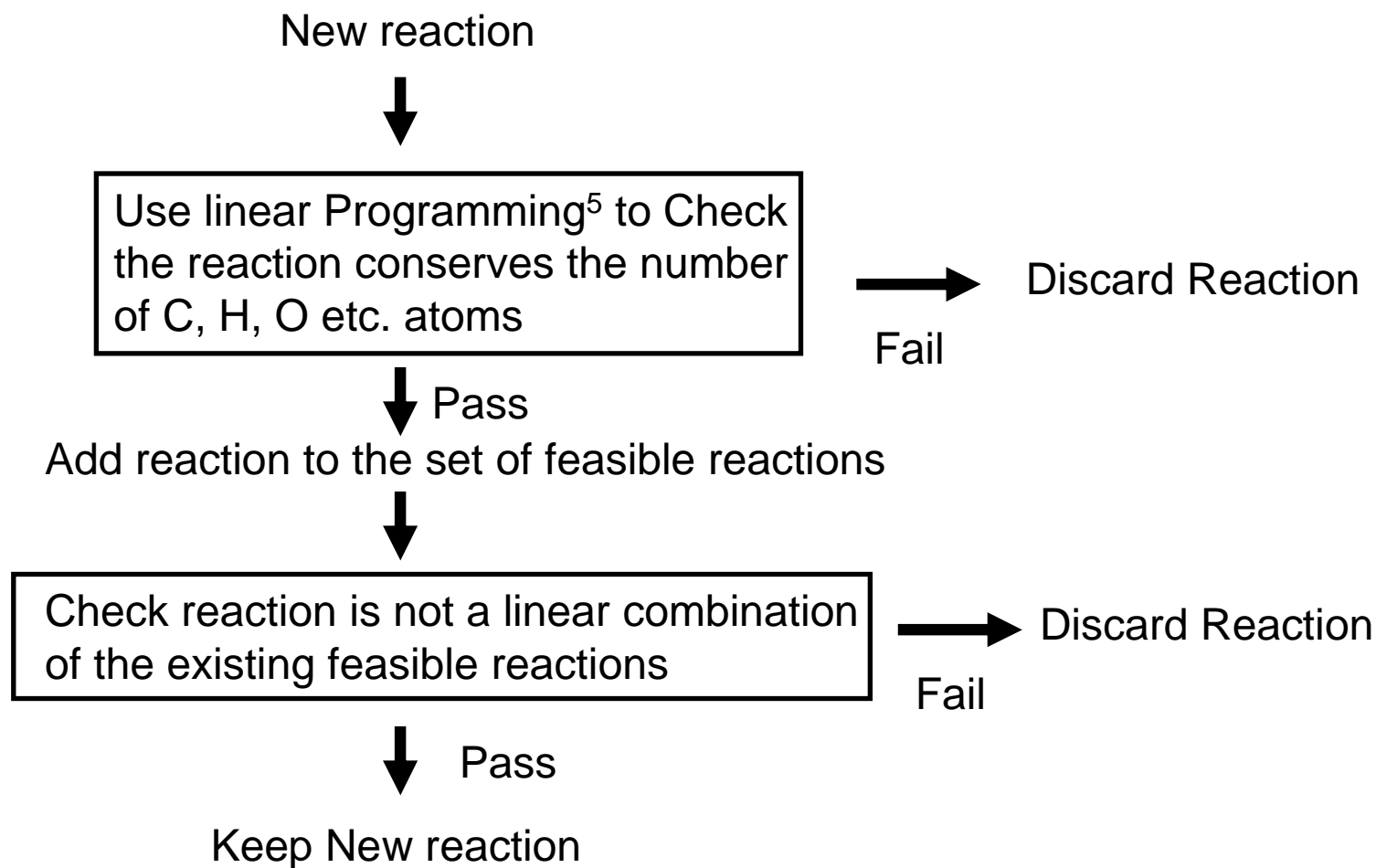
For example  $A_1 + A_2 \rightarrow$  products would not be allowed

- Do not involve stoichiometries greater than that of the known reactions

For example  $A_1 + B_1 + C_1 \rightarrow A_2 + D_2$  would not be allowed in this case

# Reaction Feasibility

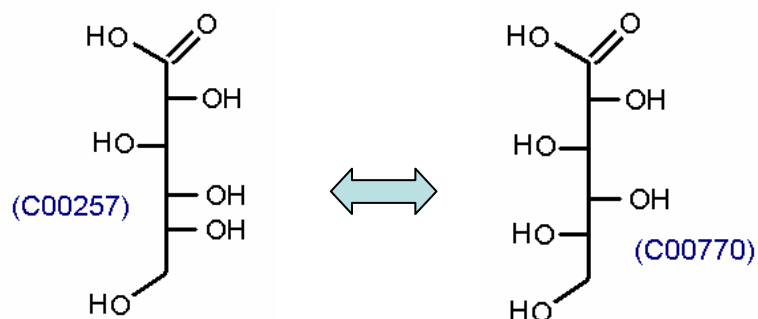
New reactions are tested for feasibility in the following ways



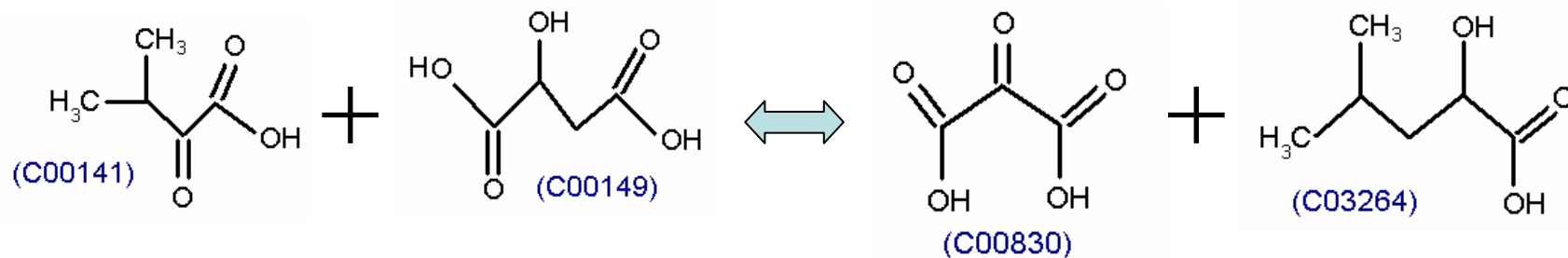
## Example: Reaction Generation for Enzyme 1.1.1.27

Reaction generation yields 241 linearly independent reactions

Including Chirality changing Reactions



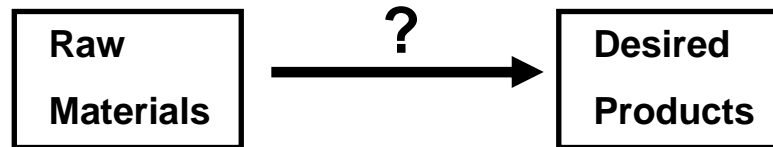
And many other new reaction involving the new binding species



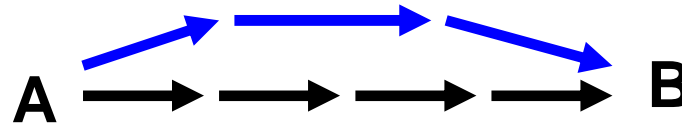
# Pathway Construction

Allows us to answer the following questions:

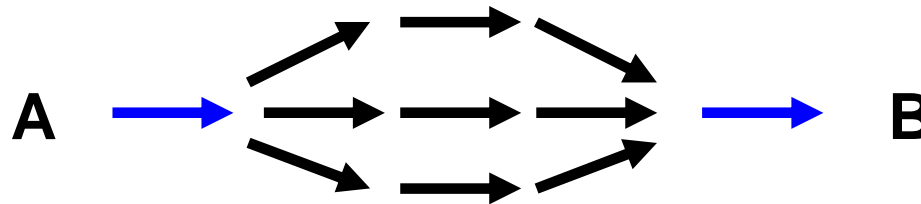
Can a Reaction network convert a set of raw materials into desired products?



Are there any alternative pathways converting A into B?

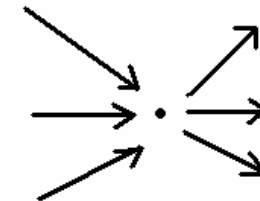


Which reactions are important for converting A into B?



Pathway construction is a complex problem for large numbers of reactions

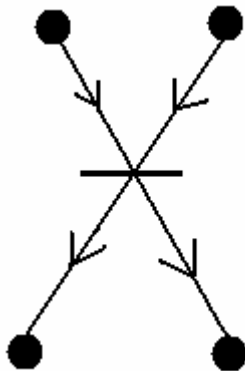
Pathway construction using P-graphs based method<sup>6</sup>



6. Seo, H., et. al., Biotechnology Letters, 23: 1551-1557, (2001).

# P-Graph Based Pathway Generation

Pathway generation is possible using an efficient implementation of p-graphs<sup>7</sup>



Dots = species/components

Bars = reactions/processes

Generates pathways according to a set of rules or axioms  
e.g. Specified Products must be produced

Using combinatorial algorithms to generate pathways

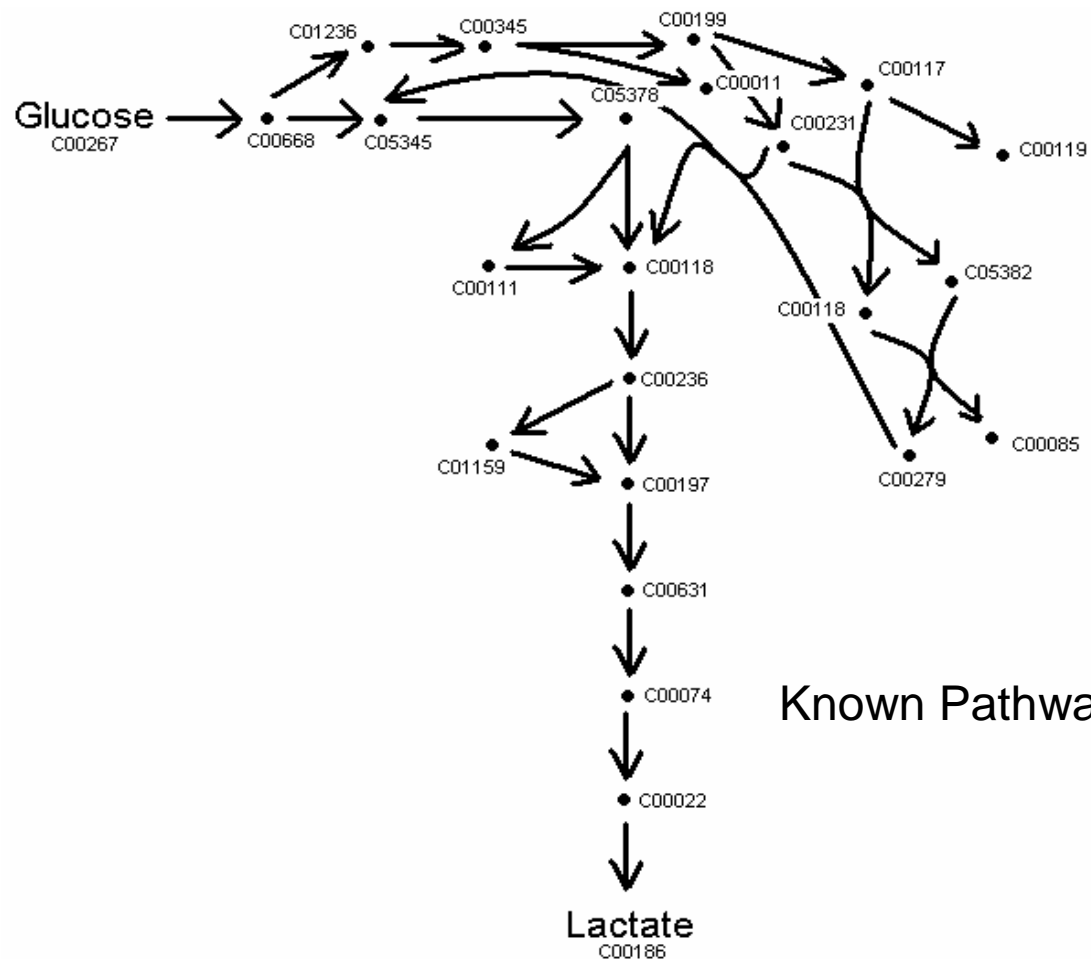
Based on branch-and-bound methods

subject to a set of rules and constraints

- Currently limited to fixed overall reactions
- Eliminates cyclic and dependent pathways
- Low memory requirements

# Human Red Blood Cell: Glycolysis

We have applied our procedure on 24 enzymes involved in the glycolysis and pentose phosphate pathways



C00267	Glucose
C00668	Glucose 6-Phosphate
C05345	Fructose 6-phosphate
C05378	Fructose 1,6 bisphosphate
C00118	Glyceraldehyde 3-phosphate
C00111	Dihydroxyacetone phosphate
C00236	1,3-Bisphospho-D-glycerate
C00197	3-Phospho-D-glycerate
C00631	2-Phospho-D-glycerate
C00074	Phosphoenolpyruvate
C00022	Pyruvate
C00186	Lactate
C00085	Fructose 6-phosphate (different chirality to C05345)
C00345	6-Phospho-D-gluconate
C00199	Ribulose 5-phosphate
C00011	Carbon dioxide
C00231	Xylulose 5-phosphate
C00117	Ribose 5-phosphate
C05382	Sedoheptulose 7-phosphate
C00279	Erythrose 4-phosphate

Known Pathways

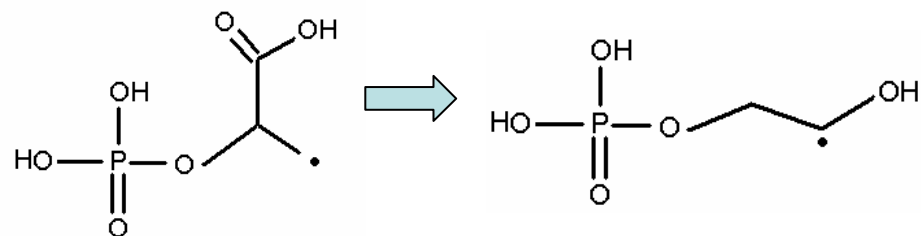


# Calculating Pharmacophores

Our procedure calculates pharmacophores for the 24 enzymes (107 overall)

Enzyme	Stoichiometry
1.1.1.27	2 → 3 and 2 → 2
1.1.1.44	2 → 4
1.1.1.49	2 → 3
1.2.1.12	3 → 3 and 3 → 3
2.2.1.1	2 → 2
2.2.1.2	2 → 2
2.7.1.1	2 → 2
2.7.1.11	2 → 2
2.7.1.2	2 → 2
2.7.1.40	2 → 2
2.7.2.3	2 → 2
2.7.6.1	2 → 2
3.1.1.31	2 → 1
3.1.3.11	2 → 2
3.1.3.13	2 → 2
3.6.1.5	3 → 3 and 2 → 2
4.1.2.13	1 → 2
4.2.1.11	1 → 2
5.1.3.1	1 → 1
5.3.1.1	1 → 1
5.3.1.6	1 → 1
5.3.1.9	1 → 1
5.4.2.1	1 → 1
5.4.2.4	2 → 2 and 1 → 1

## Enzyme 5.3.1.9

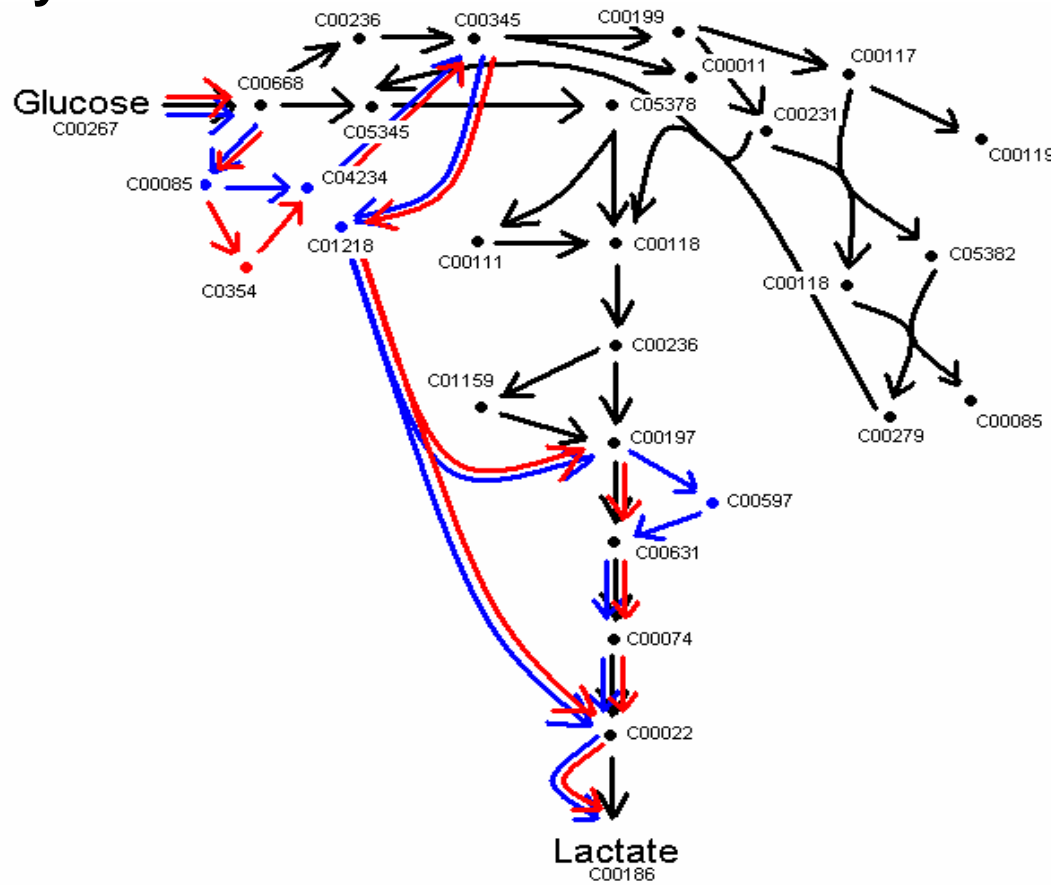


# Binding Species and Reactions

Enzyme	Binding Species		Reactions	
	Known	Calculated	Known	Calculated
1.1.1.27	9	254	3	241
1.1.1.44	6	9	1	5
1.1.1.49	6	13	2	10
1.2.1.12	12	17	3	2
2.2.1.1	9	237	6	222
2.2.1.2	4	10	1	7
2.7.1.1	24	825	22	789
2.7.1.11	16	56	13	41
2.7.1.2	8	35	3	31
2.7.1.40	18	46	8	35
2.7.2.3	4	5	1	2
2.7.6.1	4	5	1	2
3.1.1.31	3	3	1	1
3.1.3.11	8	49	4	33
3.1.3.13	4	10	1	3
3.6.1.5	20	312	13	297
4.1.2.13	10	434	5	180
4.2.1.11	7	2422	3	2039
5.1.3.1	2	4	1	3
5.3.1.1	2	3	1	2
5.3.1.6	2	6	1	5
5.3.1.9	6	26	4	16
5.4.2.1	4	76	2	28
5.4.2.4	7	82	3	31

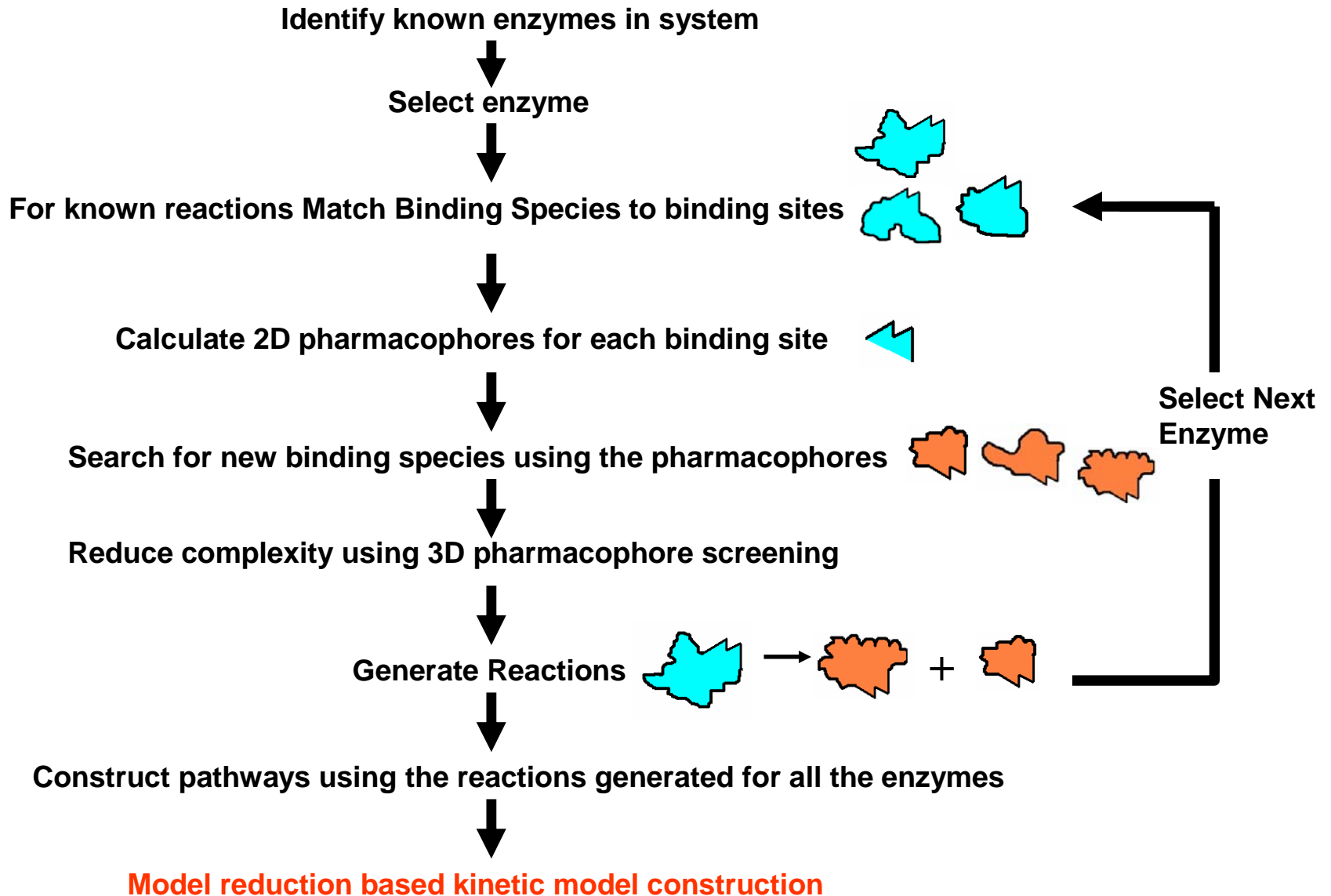
# Glycolysis: Pathway Construction

Generating new reactions and computing pathways leads to some new routes through the system



C00267	Glucose
C00668	Glucose 6-Phosphate
C05345	Fructose 6-phosphate
C05378	Fructose 1,6-bisphosphate
C00118	Glyceraldehyde 3-phosphate
C00111	Dihydroxyacetone phosphate
C00236	1,3-Bisphospho-D-glycerate
C00197	3-Phospho-D-glycerate
C00631	2-Phospho-D-glycerate
C00074	Phosphoenolpyruvate
C00022	Pyruvate
C00186	Lactate
C00085	Fructose 6-phosphate (different chirality to C05345)
C00354	Fructose 1,6-bisphosphate (different chirality to C05378)
C00345	6-Phospho-D-gluconate
C00199	Ribulose 5-phosphate
C00011	Carbon dioxide
C00231	Xylulose 5-phosphate
C00117	Ribose 5-phosphate
C00119	5-Phospho-alpha-D-ribose 1-diphosphate (PRPP)
C05382	Sedoheptulose 7-phosphate
C00279	Erythrose 4-phosphate
C00597	3-Phosphoglycerate (different chirality than C00197)
C01218	6-Phospho-2-dehydro-D-gluconate
C04234	2-Carboxy-D-arabinitol 1-phosphate

# Metabolic Network Development: A knowledge-based method for generating Biochemical reaction networks.



## Low Dimensional Manifolds

$$\frac{dc}{dt} = F(c) \quad \text{System of ODEs}$$

$$Q^T J Q = \begin{pmatrix} S_{11} & S_{12} \\ 0 & S_{22} \end{pmatrix} \quad \text{Decoupling fast and slow timescales}$$

$$S_{11} Z - Z S_{22} = -S_{12}$$

$$T = Q \left( I + \begin{pmatrix} 0 & Z \\ 0 & 0 \end{pmatrix} \right) \quad T^{-1} J T = \begin{pmatrix} S_{11} & 0 \\ 0 & S_{22} \end{pmatrix}$$

Projection onto fast/slow subspace

$$\begin{pmatrix} g_{slow} \\ g_{fast} \end{pmatrix} = T^{-1} F(T \cdot) \quad \begin{pmatrix} x_{slow} \\ x_{fast} \end{pmatrix} = T^{-1} c$$

QSSA for fast species

$$\frac{dx_{slow}}{dt} = g_{slow}(x_{slow}, x_{fast})$$

$$\frac{dx_{fast}}{dt} = 0$$

7. Maas, U., Pope, S. B., Combustion and Flame, 88:239-264, (1992).

8. Shaik, O. S., et. al., The Journal of Chemical Physics, 123, 234103, (2005)

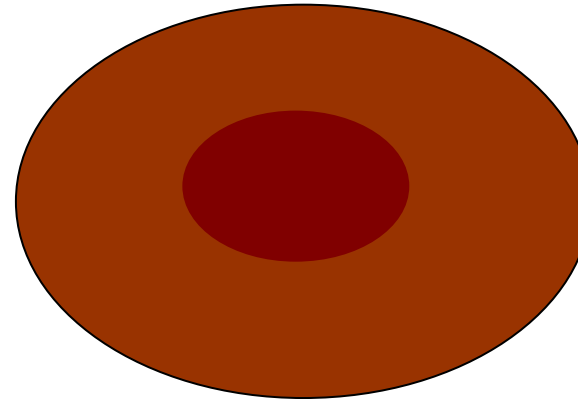
# Human Red Blood Cell

**A biochemical system representing the function of a red blood cell<sup>9</sup>**

- 49 Species
  - 5 extracellular species
  - 44 intracellular species
- 41 Reactions

A simple system

- No Nucleus
- Involves mostly Glycolysis



## Example: Human Red Blood Cell

### Applying the LDM procedure

44 ODEs and 5 algebraic equations

$$\varepsilon |g_{slow}(c_0) - g(c_{DAE})|$$

0.00074

40 ODEs and 9 algebraic equations

0.00528

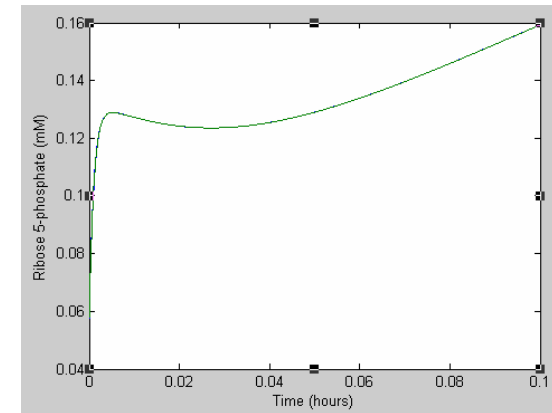
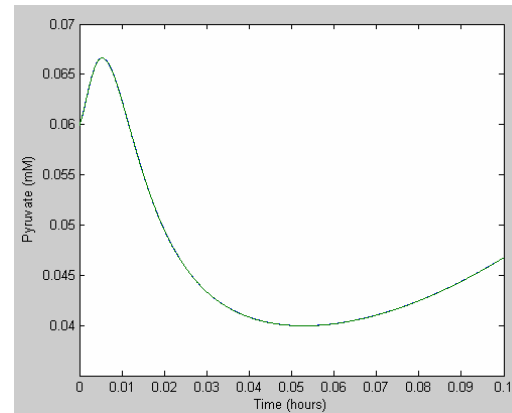
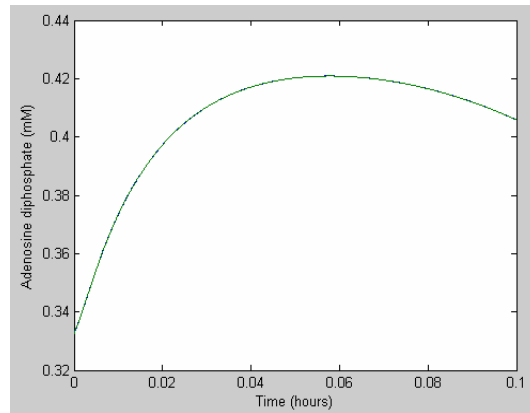
37 ODEs and 12 algebraic equations

0.0441

For this system a large number of ODEs are required

# Comparison with Full Model

37 ODEs and 12 algebraic equations



Comparison with the full kinetic model showed an almost exact match for 0.1 hours



## Species contributing to the Slow Dynamics

By analysing the projection matrix the species contributing to the fast or slow dynamics can be assessed<sup>8</sup>

$$x_{slow} = T_r^{-1} c$$

$$p_i^{slow} = \sum_{j=1}^r t_{j,i} t_j \quad p_i^{fast} = \sum_{j=r+1}^n t_{j,i} t_j$$

$$d_i^{slow} = |e_i - p_i^{slow}| \quad d_i^{fast} = |e_i - p_i^{fast}|$$

$$r_i^{slow} = \frac{d_i^{fast} \arccos d_i^{slow}}{d_i^{fast} \arccos d_i^{slow} + d_i^{slow} \arccos d_i^{fast}}$$

$$r_i^{fast} = \frac{d_i^{slow} \arccos d_i^{fast}}{d_i^{slow} \arccos d_i^{fast} + d_i^{fast} \arccos d_i^{slow}}$$

$t_j$  = column vectors of T

$t_{j,i}$  = ith component of  $t_j$

$e_i$  = unit vector (species i)

$d_i$  = distance of species i from slow/fast subspace

$p_i$  = projection of species i on the slow/fast subspace

8. Shaik, O. S., et. al., The Journal of Chemical Physics, 123, 234103, (2005)

# Species contributing to the Slow Dynamics

Analysis of  $r_i^{fast}$  and  $r_i^{slow}$

For the system with 37 slow modes and 12 fast modes shows

7 species are most associated with the fast dynamics

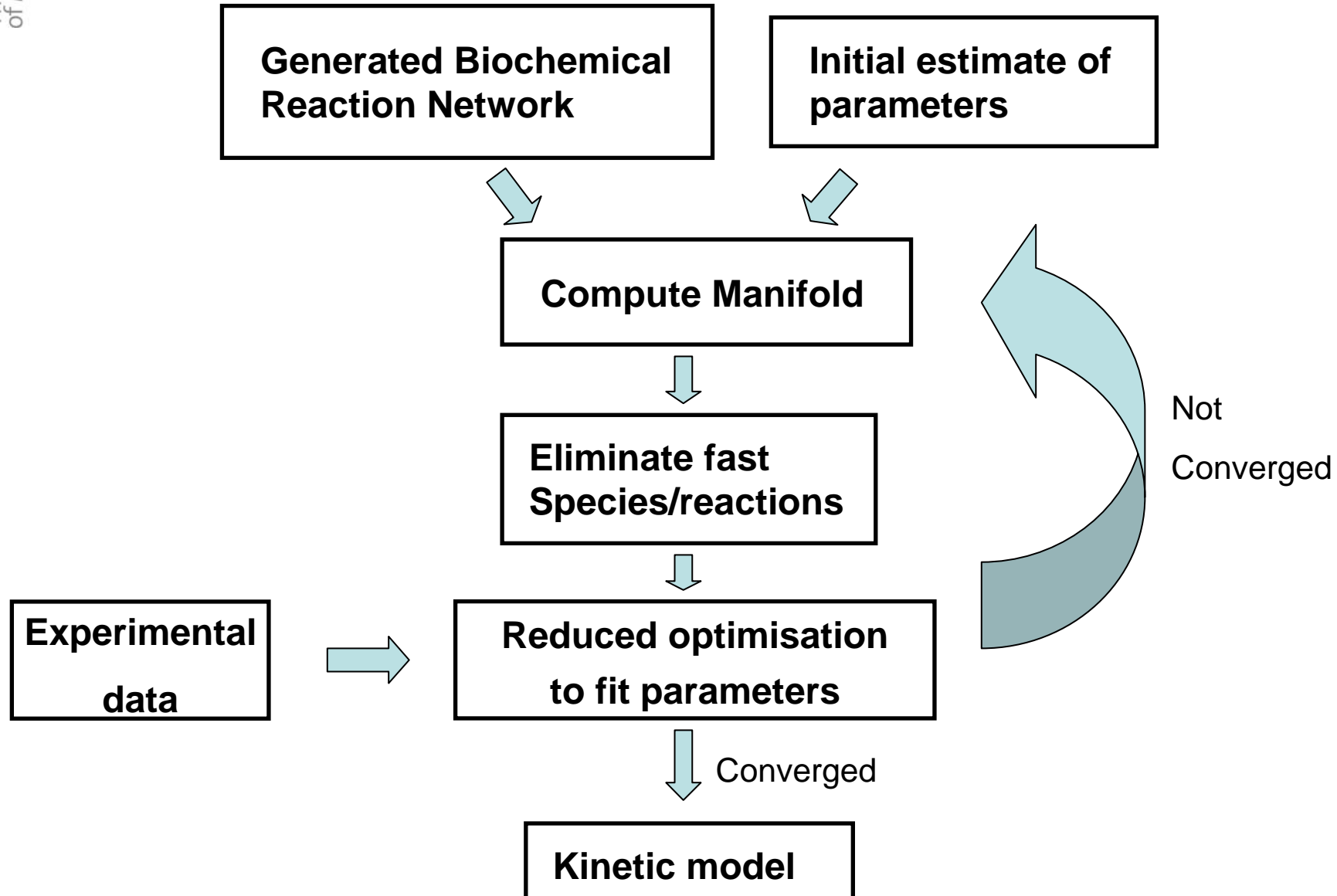
GL6P, GA3P, 1,3DPG, 2PG, PYR, INO and R1P

31 species are most associated with the slow dynamics

e.g. ATP, Glc, 2,3DPG, FDP, NADP, GSSG, LAC, etc.

And the remaining 11 species are in-between

# Parameter Estimation



# Conclusions

We have developed a complete procedure to construct a wide range of metabolic networks with minimal knowledge of the system

The Procedure:

- Finds New Binding Species
- Calculates New reactions
- Generates New pathways

We have eliminated redundant data at each step to reduce the complexity

Finding kinetic expressions and fitting parameters for large reaction networks

Should be possible if a sufficiently reduced model can be found

- Using a Low Dimensional Manifold method
- Identifying and eliminating species which do not contribute to the slow dynamics

## Acknowledgements

The financial support of the EU programme AITEKIN  
C00P-CT-2003-506667 is gratefully acknowledged

